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Nuclear chemistry is the study of reactions that involve changes in nuclear structure. The chapter on atoms, molecules, and ions introduced the basic idea of nuclear structure, that the

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nucleus of an atom is composed of protons and, with the exception of H^1 , neutrons.

Recall that the number of protons in the nucleus is called the atomic number (Z) of the element, and the sum of the number of protons and the number of neutrons is the mass number (A).

3.1: Nuclear Chemistry and Radioactive Decay -

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