

Chapter 7 Worksheet 1 Balancing Chemical Equations Answer Key

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Chapter 7 Worksheet 1 Balancing

Balancing Chemical Equations Answer Key Balance the equations below: 1) $1\text{ N}_2 + 3\text{ H}_2 \rightarrow 2\text{ NH}_3$ 2) $2\text{ KClO}_3 \rightarrow 2\text{ KCl} + 3\text{ O}_2$ 3) $2\text{ NaCl} + 1\text{ F}_2 \rightarrow 2\text{ NaF} + 1\text{ Cl}_2$ 4) $2\text{ H}_2 + 1\text{ O}_2 \rightarrow 2\text{ H}_2\text{O}$ 5) $1\text{ Pb(OH)}_2 + 2\text{ HCl} \rightarrow 2\text{ H}_2\text{O} + 1\text{ PbCl}_2$ 6) $2\text{ AlBr}_3 + 3\text{ K}_2\text{SO}_4 \rightarrow 6\text{ KBr} + 1\text{ Al}_2(\text{SO}_4)_3$ 7) $1\text{ CH}_4 + 2\text{ O}_2 \rightarrow 1\text{ CO}_2 + 2\text{ H}_2\text{O}$ 8) $1\text{ C}_3\text{H}_8 + 5\text{ O}_2 \rightarrow 3\text{ CO}_2 + 4\text{ H}_2\text{O}$ 9) $2\text{ C}_8\text{H}_{18} + 25\text{ O}_2 \rightarrow 16\text{ CO}_2 + 18\text{ H}_2\text{O}$

Chapter 7 Worksheet #1 Balancing Chemical Equations

Chapter 7 Worksheet #1 Balancing Chemical Equations

(PDF) Chapter 7 Worksheet #1 Balancing Chemical Equations ...

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Chapter 7 Worksheet 1 Balancing Chemical Equations ...

Balancing Chemical Equations - Answer Key Balance the equations below: 1) $1\text{ N}_2 + 3\text{ H}_2 \rightarrow 2\text{ NH}_3$ 2) $2\text{ KClO}_3 \rightarrow 2\text{ KCl} + 3\text{ O}_2$ 3) $2\text{ NaCl} + 1\text{ F}_2 \rightarrow 2\text{ NaF} + 1\text{ Cl}_2$ 4) $2\text{ H}_2 + 1\text{ O}_2 \rightarrow 2\text{ H}_2\text{O}$ 5) $1\text{ Pb(OH)}_2 + 2\text{ HCl} \rightarrow 2\text{ H}_2\text{O} + 1\text{ PbCl}_2$ 6) $2\text{ AlBr}_3 + 3\text{ K}_2\text{SO}_4 \rightarrow 6\text{ KBr} + 1\text{ Al}_2(\text{SO}_4)_3$ 7) $1\text{ CH}_4 + 2\text{ O}_2 \rightarrow 1\text{ CO}_2 + 2\text{ H}_2\text{O}$ 8) $1\text{ C}_3\text{H}_8 + 5\text{ O}_2 \rightarrow 3\text{ CO}_2 + 4\text{ H}_2\text{O}$ 9)

Chapter 7 Worksheet #1 Balancing Chemical Equations ...

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Chapter 7 Worksheet 1 Balancing Chemical Equations Key ...

Question: Chapter 7 Worksheet #1 Balance The Equations Below: 4) $2\text{ S} + 3\text{ O}_2 \rightarrow 2\text{ SO}_2$ 5) $7\text{ H}_2 + 1\text{ O}_2 \rightarrow 2\text{ H}_2\text{O}$ 6) $2\text{ Na} + 1\text{ Cl}_2 \rightarrow 2\text{ NaCl}$ 7) $2\text{ N}_2 + 1\text{ O}_2 \rightarrow 2\text{ N}_2\text{O}$ 8) $2\text{ Mg} + 1\text{ O}_2 \rightarrow 2\text{ MgO}$ 9) $2\text{ H}_2 + 1\text{ O}_2 \rightarrow 2\text{ H}_2\text{O}$ 10) $1\text{ Pb(OH)}_2 + 2\text{ HCl} \rightarrow 2\text{ H}_2\text{O} + 1\text{ PbCl}_2$ 11) $2\text{ Ni}_3 + 3\text{ N}_2 \rightarrow 6\text{ Ni}$ 12) $\text{Na} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{H}_2$ 13) $2\text{ HCl} + 1\text{ Ca(OH)}_2 \rightarrow \text{CaCl}_2 + 2\text{ H}_2\text{O}$ This problem has been solved! See the answer. Show transcribed image text. Expert Answer . 1. $\text{H}_2 + \text{N}_2 \rightarrow 2\text{ NH}_3$ Notice that there are 2 N on the left side but only 1 on the right so let's fix that.

Solved: Chapter 7 Worksheet #1 Balance The Equations Below ...

Chapter 7: Stoichiometry of Chemical Reactions. Search for: 7.1: Writing and Balancing Chemical Equations. Learning Objectives. By the end of this section, you will be able to: Derive chemical equations from narrative descriptions of chemical reactions.

7.1: Writing and Balancing Chemical Equations | General ...

Title: Balancing Equations Worksheet #1 - ANSWERS Author: James Last modified by: James Created Date: 11/13/2010 5:46:00 PM Other titles: Balancing Equations Worksheet #1 - ANSWERS

Balancing Equations Worksheet #1 - ANSWERS

Balancing Chemical Equations Worksheet 1 - ANSWERS 1. $2\text{ H}_2 + 1\text{ O}_2 \rightarrow 2\text{ H}_2\text{O}$ 2. $2\text{ Na} + 1\text{ Cl}_2 \rightarrow 2\text{ NaCl}$ 3. $\text{N}_2 + 1\text{ O}_2 \rightarrow 2\text{ NO}$ 4. $2\text{ Mg} + 1\text{ O}_2 \rightarrow 2\text{ MgO}$ 5. $2\text{ H}_2 + 1\text{ O}_2 \rightarrow 2\text{ H}_2\text{O}$ 6. $3\text{ Ca} + 1\text{ N}_2 \rightarrow 3\text{ Ca}_3\text{N}_2$ 7. $2\text{ Li} + 1\text{ F}_2 \rightarrow 2\text{ LiF}$ 8. $3\text{ Mg} + 1\text{ N}_2 \rightarrow 3\text{ Mg}_3\text{N}_2$ 9. $2\text{ NH}_3 + 3\text{ N}_2 + 2\text{ H}_2 \rightarrow 2\text{ H}_2\text{O}$ 10. $2\text{ HCl} \rightarrow \text{H}_2 + \text{Cl}_2$ 11. $2\text{ Ni}_3 + 3\text{ N}_2 \rightarrow 6\text{ Ni}$ 12. $2\text{ Ni}_3 + 3\text{ N}_2 \rightarrow 6\text{ Ni}$

Balancing Chemical Equations Worksheet 1

If you also get perplexed in balancing chemical equations, follow the tips for correct balancing chemical equations worksheet answers. Tip # 1: When you are trying to balance the chemical equations, you should remember that you can only change the value of coefficient in front of the element or compound, and not the subscript.

49 Balancing Chemical Equations Worksheets [with Answers]

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Balancing Chemical Equations - Answer Key Balance the equations below: 1) $1\text{ N}_2 + 3\text{ H}_2 \rightarrow 2\text{ NH}_3$ 2) $2\text{ KClO}_3 \rightarrow 2\text{ KCl} + 3\text{ O}_2$ 3) $2\text{ NaCl} + 1\text{ F}_2 \rightarrow 2\text{ NaF} + 1\text{ Cl}_2$ 4) $2\text{ H}_2 + 1\text{ O}_2 \rightarrow 2\text{ H}_2\text{O}$ 5) $1\text{ Pb(OH)}_2 + 2\text{ HCl} \rightarrow 2\text{ H}_2\text{O} + 1\text{ PbCl}_2$ 6) $2\text{ AlBr}_3 + 3\text{ K}_2\text{SO}_4 \rightarrow 6\text{ KBr} + 1\text{ Al}_2(\text{SO}_4)_3$ 7) $1\text{ CH}_4 + 2\text{ O}_2 \rightarrow 1\text{ CO}_2 + 2\text{ H}_2\text{O}$ 8) $1\text{ C}_3\text{H}_8 + 5\text{ O}_2 \rightarrow 3\text{ CO}_2 + 4\text{ H}_2\text{O}$ 9) $2\text{ C}_8\text{H}_{18} + 25\text{ O}_2 \rightarrow 16\text{ CO}_2 + 18\text{ H}_2\text{O}$ 10) $1\text{ FeCl}_3 + 3\text{ ...}$

H SO NaNO HNO Na SO

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