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# **Answers Review Chemical Formulas Compounds**

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CHAPTER 7 REVIEW Chemical Formulas  
and Chemical Compounds SECTION 2  
SHORT ANSWER Answer the following

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## Compounds

questions in the space provided. 1.  
Assign the oxidation number to the  
specified element in each of the  
following examples: 4 a. S in  $\text{H}_2\text{SO}_3$  3 6  
b. S in  $\text{MgSO}_4$  2 c. S in  $\text{K}_2\text{S}$  1 d. Cu in  
 $\text{Cu}_2\text{S}$  6 e. Cr in  $\text{Na}_2\text{CrO}_4$  5 f. N in  $\text{HNO}_3$   
3 4 g. C in  $(\text{HCO}_3)_3$  3 h. N in  $(\text{NH}_4)_2\text{SO}_4$   
2 a.

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## Compounds **7 Chemical Formulas and Chemical Compounds**

CHAPTER 7 REVIEW Chemical Formulas  
and Chemical Compounds MIXED  
REVIEW SHORT ANSWER Answer the  
following questions in the space  
provided. 1. Write formulas for the  
following compounds: CuCO<sub>3</sub> a.  
copper(II) carbonate Na<sub>2</sub>SO<sub>3</sub> b. sodium

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## Compounds

sulfite (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> c. ammonium  
phosphate SnS<sub>2</sub> d. tin(IV) sulfide HNO<sub>2</sub>  
e. nitrous acid

## **7 Chemical Formulas and Chemical Compounds**

CHAPTER 7 REVIEW Chemical Formulas  
and Chemical Compounds SECTION 2  
SHORT ANSWER Answer the following

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## Compounds

questions in the space provided. 1. Assign the oxidation number to the specified element in each of the following examples: a. S in  $\text{H}_2\text{SO}_3$  b. S in  $\text{MgSO}_4$  c. S in  $\text{K}_2\text{S}$  d. Cu in  $\text{Cu}_2\text{S}$  e. Cr in  $\text{Na}_2\text{CrO}_4$  f. N in  $\text{HNO}_3$  g. C in  $(\text{HCO}_3)$  h. N in  $(\text{NH}_4)$  2. a.

## **Chemical Formulas and Chemical**



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## Compounds - **MAFIADOC.COM**

Answers - Naming Chemical Compounds

. Name the following chemical compounds: 1) NaBr sodium bromide. 2)  $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$  calcium acetate. 3)  $\text{P}_2\text{O}_5$  diphosphorus pentoxide. 4)  $\text{Ti}(\text{SO}_4)_2$  titanium(IV) sulfate. 5)  $\text{FePO}_4$  iron(III) phosphate. 6)  $\text{K}_3\text{N}$  potassium nitride. 7)  $\text{SO}_2$  sulfur dioxide. 8)  $\text{CuOH}$

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Compounds  
copper(I) hydroxide. 9)  $Zn(NO$

## **Answers - Naming Chemical Compounds**

Review- Naming Chemical Compounds  
The following are a good mix of naming  
and formula writing problems to help  
you get some practice. Name the  
following chemical compounds:

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## **Review- Naming Chemical Compounds**

CHAPTER 7 REVIEW Chemical Formulas  
and Chemical Compounds SECTION 1  
SHORT ANSWER Answer the following  
questions in the space provided. 1. c In a  
Stock system name such as iron(III)  
sulfate, the Roman numeral tells us (a)

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## Compounds

how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

### **Modern Chemistry Chapter 7 Review Chemical Formulas And ...**

A compound is found to have a molecular mass of 90 atomic mass units and simplest formula of  $C_2H_5O$ .

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## Compounds

Question 3 A substance of phosphorus (P) and oxygen (O) is found to have a mole ratio of 0.4 moles of P for every mole of O. The simplest formula for this substance is: A.  $\text{PO}_2$  B.  $\text{P}_0.4\text{O}$  C.  $\text{P}_5\text{O}_2$  D.  $\text{P}_2\text{O}_5$

## **Chemical Formulas Practice Test Questions**

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## Compounds

MOLECULAR FORMULAS. The chemical formulas for covalent compounds are referred to as molecular formulas because these compounds exist as separate, discrete molecules. Typically, a molecular formula begins with the nonmetal that is closest to the lower left corner of the periodic table, except that hydrogen is almost never written first (H

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## Compounds

2 O is the prominent exception).

### **5.6: Covalent Compounds - Formulas and Names - Chemistry ...**

Start studying Chemical Names and Formulas Test Review. Learn vocabulary, terms, and more with flashcards, games, and other study tools. Search. ... A reaction where a compound breaks

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## Compounds

down into elements and or smaller compounds. ... animal farm studyguide answers chapters 1-4. 27 terms.  
jkhamp4u. Unit 4: Chemical Bonding Test Review. 38 terms.

## **Chemical Names and Formulas Test Review - Quizlet**

Question: Does it matter in what order



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## Compounds

you combine your elements when writing chemical formulas for ionic compounds? Explain. Ionic Bonds: Ionic compounds are formed by the strong electrostatic ...

**Solved: Does it matter in what order you combine your ...**

Review the chemical formulas for each

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## Compounds

compound listed in Data Table 5. Use the Periodic Table of Elements to determine if each element in the compound is a metal or nonmetal and record in Data Table 5. Use the information recorded in Data Tables 1 - 3 to summarize the properties of each substance and record in Data Table 5. Include the state of matter at room

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## Compounds

temperature, whether the melting and boiling points were high or low, and whether the compound was soluble or insoluble in water.

### **Solved: Review The Chemical Formulas For Each Compound Lis ...**

SHORT ANSWER 1. a. +4 b. +6 c. -2 d. +1 e. +6 f. +5 g. +4 h. -3 2. a.  $\text{SCl}_2$  b.

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## Compounds

nitrogen(IV) oxide 3. a. fluorine b. 0; F<sub>2</sub>  
4. a. tin(IV) oxide b. SnO 5. a. NO, NO<sub>2</sub>  
b. ClO 6. a. N<sub>2</sub>O<sub>3</sub> b. +3 c. The three  
oxygen atoms have oxidation states of  
-6 total, and because the algebraic sum  
of the oxidation states in a neutral  
compound must be zero, the two  
nitrogen

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## Compounds

### **CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds**

Chapter 7 - Chemical Formulas & Chemical Compounds Chapter 7 is the final chapter of the first semester. It serves as a summary of all course content discussed so far, as the focus of this chapter is on process skills rather than new concepts or theory.

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## **Chapter 7 - Chemical Formulas & Chemical Compounds - yazvac**

Near the end of class, I lead a brief class discussion about ionic and covalent bonding and ask the students to demonstrate their understanding of how to name compounds and write chemical formulas by writing their answers on the

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## Compounds

board and explaining how they arrived at those answers.

### **And the Name Is....Naming Chemical Compounds**

Naming Chemical Compounds

Worksheet - Naming Chemical

Compounds ... #143705. Naming

Compounds Handout Key - PDF

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## Compounds

#143706. The Name Says It All! (Naming Compounds and Writing Formulas ...

#143707. CHM 130LL:Formula Writing and Naming Key #143708. ... Naming Compounds Worksheet Answer Key Also Naming Ionic Pounds ... #143720. Naming ions and ionic ...

## **Naming compounds worksheet key**



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## Compounds

Question: [Review Topics] [References]  
Unshared, Or Lone, Electron Pairs Play An Important Role In Determining The Chemical And Physical Properties Of Organic Compounds. Thus, It Is Important To Know Which Atoms Carry Unshared Pairs. Use The Structural Formulas Below To Determine The Number Of Unshared Pairs At Each

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Designated Atom.

**Solved: [Review Topics]  
[References] Unshared, Or Lone, El**

...

Honors Chemistry Name Chemical  
Formula Writing Review Worksheet #7  
Period Write chemical formulas for the  
compounds in each box. The names are

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## Compounds

found by finding the intersection  
between the cations and anions.

Example: The first box is the intersection  
between the "zinc" cation and the  
"chloride" anion, so you should write

"ZnCl<sub>2</sub>", as shown. chloride acetate  
nitrate oxide nitride sulfate zinc ZnCl<sub>2</sub>

iron (II) iron (III) gallium 10) 11) 12)

gallium GaCl<sub>3</sub> Ga(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>3</sub> Ga<sub>2</sub>O<sub>3</sub> GaN

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Compounds  
10) 11) 12) silver ...

## **Hudson City Schools / Homepage**

e) Ask questions about chemical names to identify patterns in IUPAC nomenclature in order to predict chemical names for ionic (binary and ternary), acidic, and inorganic covalent compounds. f) Develop and use bonding

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## Compounds

models to predict chemical formulas including ionic (binary and ternary), acidic, and inorganic covalent compounds.

### **Unit 3 - MHS Honors Chemistry**

Formulas \*\* start studying chemical names and formulas test review learn vocabulary terms and more with

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## Compounds

flashcards games and other study tools  
chapter 7 review chemical formulas and  
chemical compounds section 1 short  
answer answer the following questions  
in the space provided 1 c in a stock

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Compounds

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